# metal-organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

## Shao-Wen Chen, Han-Dong Yin\* and Da-Qi Wang

College of Chemistry and Chemical Engineering, Liaocheng University, Shandong 252059, People's Republic of China

Correspondence e-mail: handongyin@lctu.edu.cn

#### **Key indicators**

Single-crystal X-ray study T = 298 KMean  $\sigma$ (C–C) = 0.006 Å R factor = 0.031 wR factor = 0.083 Data-to-parameter ratio = 14.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

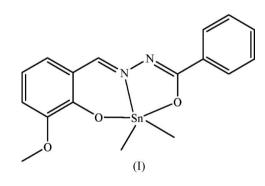
# (3-Methoxy-2-oxidobenzaldehyde benzoylhydrazonato)dimethyltin(IV)

In the title complex,  $[Sn(CH_3)_2(C_{15}H_{12}N_2O_3)]$ , the Sn atom is found in a distorted trigonal-bipyramidal geometry with Sn – O distances of 2.131 (3) and 2.178 (3) Å. Intermolecular Sn···O contacts stabilize the crystal packing.

Received 17 May 2006 Accepted 6 June 2006

## Comment

The title molecule, (I) (Fig. 1), is a monomer in which the Sn atom is five-coordinated by two O atoms, two C atoms and one N atom (Table 1). The Schiff base is coordinated to the Sn atom and the two deprotonated O atoms. The Sn1-N2 distance is 2.188 (3) Å, which is close to the sum of the nonpolar covalent radii (2.15 Å; Sanderson, 1967), indicating a strong tin-nitrogen interaction. The distorted trigonal-bipyramidal geometry around the Sn atom is the result of the strain imposed by the tridentate Schiff base ligand, and from the constraints imposed by the five- and six-membered rings Sn-N-N-C-O and Sn-N-C-C-C-O. The dihedral angle between these two rings is 0.7 (2)°, indicating their coplanarity. The distorted geometry can be seen from departures of bonding angles from the values characteristic for trigonalbipvramidal coordination (Table 1): the C16-Sn1-C17 angle should be  $180^{\circ}$ , but its value is  $140.97 (16)^{\circ}$ . The structure of the title compound is very similar to that of a compound already reported (Camacho-Camacho et al., 1998).



The crystal packing of the title complex displays a weak  $Sn \cdots O(1 - x, 2 - y, 1 - z)$  interaction [3.095 (2) Å] which stabilizes the crystal structure.

## Experimental

The reaction was carried out under a nitrogen atmosphere with use of the standed Schlenk technique. The Schiff base (0.2692 g, 1.0 mmol) was added to a mixture of ethanol and benzene (1:3, 30 ml) with sodium ethoxide (0.068 g, 1.0 mmol); the mixture was stirred for 0.5 h and  $(CH_3)_2SnCl_2$  (0.2196 g, 1.0 mmol) was added; the mixture was stirred for 7 h under reflux. After cooling to room temperature, the

© 2006 International Union of Crystallography All rights reserved solution was filtered and evaporated to dryness. The title solid was recrystallized from dichloromethane–hexane (1:1  $\nu/\nu$ ; m. p. 486–487 K). Analysis C<sub>17</sub>H<sub>18</sub>N<sub>2</sub>O<sub>3</sub>Sn: C 48.95, H 4.35, N 6.71%; found: C 48.65, N 4.12, N 6.54%.

Z = 8

 $D_x = 1.658 \text{ Mg m}^{-3}$ 

 $0.49 \times 0.43 \times 0.30$  mm

8516 measured reflections 2947 independent reflections

 $w = 1/[\sigma^2(F_0^2) + (0.042P)^2]$ 

+ 2.9384*P*] where  $P = (F_0^2 + 2F_c^2)/3$ 

 $\Delta \rho_{\rm min} = -0.50 \text{ e} \text{ Å}^{-3}$ 

 $(\Delta/\sigma)_{\text{max}} = 0.002$  $\Delta\rho_{\text{max}} = 0.48 \text{ e } \text{\AA}^{-3}$ 

2381 reflections with  $I > 2\sigma(I)$ 

Mo  $K\alpha$  radiation  $\mu = 1.55 \text{ mm}^{-1}$ 

T = 298 (2) K

Block, orange

 $R_{\rm int}=0.040$ 

 $\theta_{\rm max} = 25.0^{\circ}$ 

#### Crystal data

 $[Sn(CH_3)_2(C_{15}H_{12}N_2O_3)]$   $M_r = 417.02$ Monoclinic, C2/c a = 26.477 (4) Å b = 9.5444 (16) Å c = 13.594 (2) Å  $\beta = 103.475$  (2)° V = 3340.8 (9) Å<sup>3</sup>

#### Data collection

Siemens SMART CCD areadetector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  $T_{\min} = 0.518, T_{\max} = 0.654$ 

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.031$   $wR(F^2) = 0.083$  S = 1.022947 reflections 209 parameters H-atom parameters constrained

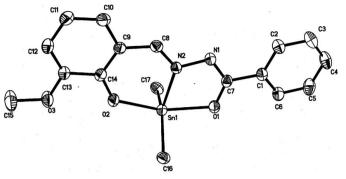
#### Table 1

Selected geometric parameters (Å, °).

Sn1-C16	2.099 (4)	Sn1-N2	2.188 (3)
Sn1-C17	2.102 (4)	N1-C7	1.306 (5)
Sn1-O2	2.131 (3)	N1-N2	1.405 (4)
Sn1-O1	2.178 (3)	N2-C8	1.285 (5)
C16-Sn1-C17	140.97 (16)	O2-Sn1-N2	83.39 (11)
C17-Sn1-O2	97.46 (14)	O1-Sn1-N2	72.01 (11)
O2-Sn1-O1	154.34 (11)		

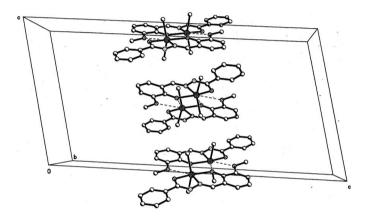
All H atoms were placed geometrically and treated as riding on their parent atoms, with aromatic C–H distances of 0.93 Å, and methyl C–H distances of 0.96 Å. The  $U_{\rm iso}({\rm H})$  values were set at  $1.5U_{\rm eq}({\rm C})$  for the methyl H atoms and at  $1.2U_{\rm eq}({\rm C})$  for the other H atoms.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.



#### Figure 1

The structure of the title complex and the atom-numbering, with displacement ellipsoids shown at the 30% probability level. H atoms have been omitted.



#### Figure 2

Partial crystal packing of the title complex, with the intermolecular  $Sn \cdots O$  interactions (dashed lines). H atoms have been omitted.

We acknowledge the financial support of the Shandong Province Science Foundation and the State Key Laboratory of Crystalline Materials, Shandong University, People's Republic of China.

### References

- Camacho-Camacho, C., Tlahnext, H., Nöth, H. & Contreras, R. (1998). Heteroat. Chem. 9, 321–326.
- Sanderson, R. T. (1967). Inorganic Chemistry, p. 74. New York: Reinhold.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997a). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.